ramic

Direct β -C(sp³)–H Functionalization of Aliphatic Amines to α,β -Unsaturated Imines, Aldehydes, and Chromenes

Sumana Mandal,† Sujit Mahato,† and Chandan K. Jana*

Department of Che[mis](#page-3-0)try, Indian Instit[ut](#page-3-0)e of Technology Guwahati, [G](#page-3-0)uwahati 781039, Assam, India

S Supporting Information

[AB](#page-2-0)STRACT: [A metal-free](#page-2-0) method for direct β -C(sp³)-H functionalization of aliphatic amine was developed. The method is based on a reaction that yields enamine directly from the corresponding aliphatic amine, which otherwise requires the aid of metallic reagent and/or external oxidant. The reaction is operationally simple, general, and highly efficient in functionalizing both cyclic and acyclic amines. Structurally diverse

unsaturated imines were obtained from N-heterocycles, while acyclic amines provided 2-alkyl cinnamaldehyde and benzopyran derivatives with excellent E/Z-selectivity.

Aliphatic amines with varied functionalization are key
structural motifs of many biologically important molecules of natural and synthetic origin.¹ In this regard, α , β -unsaturated imines are of particular interest because they can participate in different reactions, producing [v](#page-3-0)arious functionalized amines.² Moreover, different bioactive molecules are decorated with this moiety (Scheme 1A). The relevant examples include a class [of](#page-3-0) natural products having an $α, β$ -unsaturated pyrroline unit, lanopylin B1 and their derivatives, acting as a human lanosterol

Scheme 1. (A) Natural and Synthetic α , β -Unsaturated Imines; (B) $β$ -C(sp³)−H Functionalization of Aliphatic Amine; (C) First Result for Metal- and Oxidant-Free β - $C(sp^3)$ –H Functionalization

synthase inhibitor. 3 A similar structural motif was found in the alkaloid isolated from the venom of the Costa Rican ant.⁴ F[u](#page-3-0)rther, α , β -unsaturated pyrrolines are direct precursors of the β -substitute[d](#page-3-0) pyrrole derivatives, and thus, these can be used for the syntheses of pyrrole-based natural products.⁵ In addition to biological significance, the α , β -unsaturated pyrroline moiety is also very important in material chemistry in t[he](#page-3-0) context of the development of biomimetic molecular photoswitches.⁶

Various synthetic methodologies were developed for the preparation of functionalized aliphatic amines.⁷ Direct [C](#page-3-0)−H functionalization of suitable saturated amines has become the method of choice in synthesis.⁸ This strategy pr[ov](#page-3-0)ides products with desired complexity, in a single step, avoiding the additional $step(s)$ to preactivate or pref[un](#page-3-0)ctionalize the substrates. In this context, α-C−H functionalization of amine has dominated the field.^{7,8} On the other hand, although indirect pathways are known for the preparation of β -functionalized amine,⁹ reports on [dire](#page-3-0)ct functionalization of more challenging β -C(sp³)-H bonds are very few.10 Reactions primarily using meta[l-](#page-3-0) and/or oxidant-based reagent/catalyst were employed to achieve direct β -C(sp³)–H functi[on](#page-3-0)alization of amines (Scheme 1B). Moreover, removal of the N-protecting group, which is essential in metal-mediated C−H functionalization, is required for further application of functionalized amines. The known reaction on β -C−H functionalization without metal and oxidant necessitates the use of suitably prefunctionalized amine as substrate. 11 Development of novel methodology that operates under metaland oxidant-free conditions for direct β -functionalization [of](#page-3-0) aliphatic amine is desirable. Herein, we report a metal- and oxidant-free novel method for direct β -C(sp³)–H functionalization of aliphatic amines. Cyclic aliphatic secondary amines produced α , β -unsaturated imines, while α , β -unsaturated aldehydes and chromenes were obtained from acyclic amines.

Received: June 16, 2015 Published: July 23, 2015

During the development of direct α -C−H arylation of amine, 12 we reacted pyrrolidine in the presence of 9-fluorenone and p-nitrophenol in refluxing benzene (Scheme 1C). Howe[ver](#page-3-0), we observed that, instead of the desired C−H arylated product, compound 1 was formed with [26% yield.](#page-0-0) We realized that the reaction will provide the opportunity to achieve direct β -C−H functionalization of aliphatic amines under simple reaction conditions without using metal- or oxidant-based reagents or catalyst. Furthermore, the method will have potential to provide privileged structures in a single step without using preoxidized or prefunctionalized substrates. Therefore, we investigated the reaction further. Reaction conditions varying the additives, solvents, reaction times, and temperatures were examined to optimize the reaction of pchlorobenzaldehyde with pyrrolidine (see the Supporting Information, Table S1). The best yield (94%) of desired product 2a was obtained in a reaction of ald[ehyde with](#page-2-0) [pyrrolidine \(](#page-2-0)4 equiv) in the presence of 3,5-dinitrobenzoic acid (DNBA, 0.6 equiv) in refluxing xylene. The isolated product was found to have the double bond exclusively with Econfiguration.

The optimized reaction conditions were employed to investigate the substrate scope of the reaction. Various aldehydes containing a wide range of functional groups were reacted with different N-heterocycles providing structurally diverse α , β -unsaturated imines 2a−t (Scheme 2). Aldehydes

Scheme 2. Scope in Synthesis of α , β -Unsaturated Imines

with or without an electron-donating or electron-withdrawing substituent in the aryl moiety were equally efficient in providing corresponding β -functionalized amines. Moreover, the reactions worked well with aryl aldehydes having substituents irrespective of their position ortho, meta, and para to aldehyde moiety. Heteroaromatic aldehyde also provided the desired imine 2p with excellent yield (88%). Lower yields of imine derivatives (2q−s) were obtained by functionalizing piper $idine.¹³$ Similarly, cinnamaldehyde derivative was also employed successfully to obtain conjugated imine 2t with very good yield. To o[ur](#page-3-0) surprise, dehomologation occurred, producing pyrroline

2m with 76% yield, while cinnamaldehyde was reacted with pyrrolidine under the standard reaction conditions (Scheme 3).

Retro-aldol reaction is proposed to rationalize this observa- χ tion.¹⁴ In the presence of excess pyrrolidine, cinnamaldehyde probably participated in conjugate addition to provide β -amino imin[ium](#page-3-0) ion derivative 3. Subsequently, 3 underwent retroaldol reaction forming benzaldehyde or its corresponding iminium ion, which reacted with pyrrolidine under standard reaction conditions providing pyrroline 2m.

The reaction with 2-substituted pyrrolidine was investigated next. Accordingly, L-proline was reacted with p-chlorobenzaldehyde under optimized reaction conditions. However, decarboxylation occurred and compound 2a (10%) was isolated along with other undesired derivatives (Scheme S1, eq 1). On the other hand, two regioisomeric conjugated imines 4a and 4b were obtained from 2-methyl pyrrolidin[e \(Scheme](#page-2-0) 3).

With the success in functionalizing aliphatic N-heterocycles, we turned our attention to acyclic saturated secondary amines as the potential substrates for β -C−H functionalization. We set the first reaction of N_jN -dibutylamine with p-chlorobenzaldehyde under the reaction conditions optimized for N-heterocycles. However, the homologated unsaturated aldehyde 6a was isolated with 50% yield instead of desired unsaturated imine 5 (Scheme 4 and Scheme S2). Unsaturated aldehydes were widely used in organic synthesis for the preparation of bioactive [natural and](#page-2-0) un[natural com](#page-2-0)pounds.¹⁵ Specially, 2- alkylsubstituted cinnamaldehyde derivatives find direct application in the perfume and cosmetic indust[ry.](#page-3-0)¹⁶ Syntheses of these compounds, primarily via cross-aldol reaction, remained inefficient due to the associated unde[sire](#page-3-0)d self-condensation and polymerization reaction.¹⁷ These complications could be potentially avoided during their syntheses via this operationally simple method utilizing a[min](#page-3-0)e as the formal aldol donor. Therefore, we looked further to optimize (Table S2) and investigate the scope of the reaction. Different dialkylamines were reacted with various aldehydes provi[ding struct](#page-2-0)urally diverse 2-alkylated cinnamaldehyde derivatives 6a−q (Scheme 4). Many of them can be considered as potential aroma substances for use in the fragrance industry. In particul[ar, hexyl](#page-2-0) [ci](#page-2-0)nnamal (6l) is a natural aroma found in the essential oil of chamomile and used in perfume. This compound was prepared from benzaldehyde and dioctylamine in a single step without using metal- or oxidant-based reagents. Conjugated aldehydes were found to have thermodynamically more stable E-geometry in the double bond. However, mixtures of E- and Z-isomers were found for aldehyde 6g and dienals 6p and 6q. Interestingly, coumarin and/or chromene-2-ol derivatives

Scheme 4. Scope in Synthesis of 2-Alkylated Cinnamaldehyde

(7a−d) with a Z-double bond were obtained on reaction with 2-hydroxyaryl aldehydes. As these compounds belong to an important class of heterocycles having biological importance,¹⁸ salicylaldehyde derivatives were reacted with different dialkyl amines to obtain structurally diverse benzopyran derivativ[es](#page-3-0) (7a−d). 2-Alkylcoumarins were obtained from the reaction of salicylaldehyde, while o-vanillin gave chromene-2-ol as the major product.

On the basis of the experimental findings, a mechanistic proposal for the metal- and oxidant-free direct $β$ -C−H functionalization of secondary amine is presented in Scheme 5. Aldehyde 8 condensed with secondary aliphatic amine

Scheme 5. Mechanistic Proposal for Direct $β$ -C(sp 3)–H Functionalization

providing iminium ion 9, which then rearranged to isomeric iminium ion 10. Nucleophilic substitution reaction occurred at the benzylic or allylic position of 10 to release imine 11 and benzyl alcohol 12 or benzyl amine 13. Subsequent reaction of enamine, formed from imine 11, with aldehyde 8 provided the desired α , β -unsaturated imine 2 via intermediate alcohol 14. For acyclic amine, the corresponding unstable imine 5 underwent hydrolysis providing conjugated aldehyde 6. For salicylaldehyde-based substrates, the hydroxy group promoted intramolecular cyclization providing chromene-2-ol that underwent subsequent oxidation to coumarin derivatives 7. Ring closing occurred either via thermal 6π-electrocyclization of oquinone methide 15 or through intramolecular hemiacetalization (Scheme S2).19 Attempts were made to isolate benzyl amine 13 as the support for the mechanistic proposal. Tertiary amine 6aa corresp[on](#page-3-0)ding to 13 was isolated along with the conjugated aldehyde 6a (see Table S2). We thought that benzyl pyrrolidine 13 and benzyl alcohol 12, which were not detected, probably reacted further, giving desired imine 2. As a test for this, we found that, under the same reaction conditions, benzyl alcohol and benzyl pyrrolidine reacted separately in the presence of pyrrolidine to provide the desired imine 2m (Scheme S1, eq 2 and 4). A considerable reduction of yield (to 46%) of 2a was observed in a reaction performed under argon atmosphere (Table S1, entry 17). Therefore, the direct reaction of imine 11, formed via aerial oxidation of pyrrolidine, 20 with aldehyde also contributed to the yield of 2 (Scheme S2). However, the experiment in the presence of radical i[nh](#page-3-0)ibitor (BHT) did not show significant decrease in the yields of the desired product (Table S1).

Functionalized heterocycles obtained via this method can be further derivatized in a varied way to provide valuable products. For example, the imine functionality of 2 was reduced to afford allylic amines 16a and16b (Scheme 6). Addition of

benzylmagnesium chloride furnished α , β -difunctionalized amines 17a and 17b with very good yields. Furthermore, valuable 3-substituted pyrrole derivatives 18a and 18b were obtained via base-mediated aromatization of the corresponding β -functionalized imines.

We have discovered a novel method for the direct β -C−H functionalization of aliphatic amines. Enamines were formed in situ directly from corresponding aliphatic amines without the aid of metal-based reagents and external oxidant. The method is general as it functionalizes both cyclic and acyclic saturated amines. Cyclic aliphatic amines provided biologically as well as synthetically important α , β -unsaturated imines with good to excellent yields and excellent E-selectivity. As the formal aldol donor, acyclic amine reacted with aldehyde under simple reaction conditions producing a series of 2-alkylcinnamaldehyde and chromen-2-ol/-one derivatives including natural aromas.

■ ASSOCIATED CONTENT

S Supporting Information

Tables, additional schemes, experimental details, spectral data of all compounds, and X-ray data for compound 1 (CIF). The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.5b01744.

Organic Letters
■ AUTHOR INFORMATION

Corresponding Author

*E-mail: ckjana@iitg.ernet.in.

Author Contributions

† S.M. and S.M. contributed equally.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

C.K.J. acknowledges SERB, New Delhi, for funding.

■ REFERENCES

(1) For selected reviews, see: (a) Felpin, F.-X.; Lebreton, J. Tetrahedron 2004, 60, 10127. (b) Lourenco, A. M.; Máximo, P.; Ferreira, L. M.; Pereira, M. M. A. Studies in Natural Products Chemistry: Bioactive Natural Products (Part H); Rahman, A., Ed.; Elsevier: Amsterdam, 2002; Vol. 27, p 233. (c) Liddell, J. R. Nat. Prod. Rep. 2002, 19, 773. For selected reports, see: (d) Wagner, F. F.; Comins, D. L. Tetrahedron 2007, 63, 8065. (e) Dieter, R. K.; Chen, N.; Watson, R. T. Tetrahedron 2005, 61, 3221.

(2) (a) Jiang, X.; Wang, R. Chem. Rev. 2013, 113, 5515. (b) Duttwyler, S.; Lu, C.; Rheingold, A. L.; Bergman, R. G.; Ellman, J. A. J. Am. Chem. Soc. 2012, 134, 4064.

(3) (a) Sakano, Y.; Shibuya, M.; Matsumoto, A.; Takahashi, Y.; Tomoda, H.; O̅mura, S.; Ebizuka, Y. J. Antibiot. 2003, 56, 817.

(4) (a) Jones, T. H.; DeVries, P. J.; Escoubas, P. J. Chem. Ecol. 1991, 17, 2507.

(5) (a) Liao, J.-Y.; Shao, P.-L.; Zhao, Y. J. Am. Chem. Soc. 2015, 137, 628. (b) Yu, Y.; Wang, C.; He, X.; Yao, X.; Zu, L. Org. Lett. 2014, 16, 3580. (c) Oda, M.; Fujiwara, Y.; Shoji, T.; Abe, T.; Kuroda, S. Heterocycles 2012, 85, 1187. (d) Wittig, G.; Hesse, A. Liebigs Ann. Chem. 1975, 1975, 1831.

(6) (a) Blanco-Lomas, M.; Samanta, S.; Campos, P. J.; Woolley, G. A.; Sampedro, D. J. Am. Chem. Soc. 2012, 134, 6960. (b) Léonard, J.; Schapiro, I.; Briand, J.; Fusi, S.; Paccani, R. R.; Olivucci, M.; Haacke, S. Chem. - Eur. J. 2012, 18, 15296. (c) Rossi Paccani, R. R.; Donati, D.; Fusi, S.; Latterini, L.; Farina, G.; Zanirato, V.; Olivucci, M. J. Org. Chem. 2012, 77, 1738. (d) Lumento, F.; Zanirato, V.; Fusi, S.; Busi, E.; Latterini, L.; Elisei, F.; Sinicropi, A.; Andruniów, T.; Ferré, N.; Basosi, R.; Olivucci, M. Angew. Chem., Int. Ed. 2007, 46, 414.

(7) For selected reviews on amine functionalization, see: (a) Seidel, D. Acc. Chem. Res. 2015, 48, 317. (b) Haibach, M. C.; Seidel, D. Angew. Chem., Int. Ed. 2014, 53, 5010. (c) Peng, B.; Maulide, N. Chem. - Eur. J. 2013, 19, 13274. (d) Shi, L.; Xia, W. Chem. Soc. Rev. 2012, 41, 7687. (f) Mitchell, E. A.; Peschiulli, A.; Lefevre, N.; Meerpoel, L.; Maes, B. U. W. Chem. - Eur. J. 2012, 18, 10092. (e) Jones, K. M.; Klussmann, M. Synlett 2012, 2012, 159. (f) Doye, S. Angew. Chem., Int. Ed. 2001, 40, 3351.

(8) For selected recent reports on α -functionalization of amine, see: (a) Xie, Z.; Liu, L.; Chen, W.; Zheng, H.; Xu, Q.; Yuan, H.; Lou, H. Angew. Chem., Int. Ed. 2014, 53, 3904. (b) Mori, K.; Kurihara, K.; Yabe, S.; Yamanaka, M.; Akiyama, T. J. Am. Chem. Soc. 2014, 136, 3744. (c) Mahato, S.; Haque, M. A.; Dwari, S.; Jana, C. K. RSC Adv. 2014, 4, 46214. (d) Richers, M. T.; Breugst, M.; Platonova, A. Y.; Ullrich, A.; Dieckmann, A.; Houk, K. N.; Seidel, D. J. Am. Chem. Soc. 2014, 136, 6123. (e) Mahato, S.; Haldar, S.; Jana, C. K. Chem. Commun. 2014, 50, 332. (f) Chen, W.; Wilde, R. G.; Seidel, D. Org. Lett. 2014, 16, 730. (g) Lin, W.; Cao, T.; Fan, W.; Han, Y.; Kuang, J.; Luo, H.; Miao, B.; Tang, X.; Yu, Q.; Yuan, W.; Zhang, J.; Zhu, C.; Ma, S. Angew. Chem., Int. Ed. 2014, 53, 277. (h) He, Y.-P.; Wu, H.; Chen, D.-F.; Yu, J.; Gong, L.-Z. Chem. - Eur. J. 2013, 19, 5232. (i) Zheng, Q.-H.; Meng, W.; Jiang, G.-J.; Yu, Z.-X. Org. Lett. 2013, 15, 5928. (j) Das, D.; Seidel, D. Org. Lett. 2013, 15, 4358. (k) Zhu, Y.; Zhao, H.; Wei, Y. Synthesis 2013, 45, 952. (l) Xia, X.-F.; Zhang, L.-L.; Song, X.-R.; Niu, Y.-N.; Liu, X.-Y.; Liang, Y.-M. Chem. Commun. 2013, 49, 1410. (m) Das, D.; Sun, A. X.; Seidel, D. Angew. Chem., Int. Ed. 2013, 52, 3765. (n) Chen, D.-

F.; Han, Z.-Y.; He, Y.-P.; Yu, J.; Gong, L.-Z. Angew. Chem., Int. Ed. 2012, 51, 12307. (o) Jurberg, I. D.; Peng, B.; Wö stefeld, E.; Wasserloos, M.; Maulide, N. Angew. Chem., Int. Ed. 2012, 51, 1950. (p) Boess, E.; Schmitz, C.; Klussmann, M. J. Am. Chem. Soc. 2012, 134, 5317. (q) Alagiri, K.; Devadig, P.; Prabhu, K. R. Chem. - Eur. J. 2012, 18, 5160 and reports cited in ref 7.

(9) (a) Blanco-Lomas, M.; Campos, P. J.; Sampedro, D. Eur. J. Org. Chem. 2012, 2012, 6328. (b) O'Reilly, M. C.; Lindsley, C. W. Org. Lett. 2012, 14, 2910. (c) Shi, Z.; Yu, P.; Loh, T.-P.; Zhong, G. Angew. Chem., Int. Ed. 2012, 51, 7825. (d) Barber, D. M.; Sanganee, H. J.; Dixon, D. J. Org. Lett. 2012, 14, 5290. (e) Kim, M.; Park, Y.; Jeong, B.- S.; Park, H.; Jew, S. Org. Lett. 2010, 12, 2826. (f) Nomura, Y.; Bando, T.; Takeuchi, Y.; Tomoda, S. Bull. Chem. Soc. Jpn. 1983, 56, 3199.

(10) (a) Takasu, N.; Oisaki, K.; Kanai, M. Org. Lett. 2013, 15, 1918. (b) Boudiar, T.; Sahli, Z.; Sundararaju, B.; Achard, M.; Kabouche, Z.; Doucet, H.; Bruneau, C. J. Org. Chem. 2012, 77, 3674. (c) Yuan, K.; Jiang, F.; Sahli, Z.; Achard, M.; Roisnel, T.; Bruneau, C. Angew. Chem., Int. Ed. 2012, 51, 8876. (d) Sundararaju, B.; Achard, M.; Sharma, G. V. M; Bruneau, C. J. Am. Chem. Soc. 2011, 133, 10340. (e) Xia, X.-F.; Shu, X.-Z.; Ji, K.-G.; Shaukat, A.; Liu, X.-Y.; Liang, Y.-M. J. Org. Chem. 2011, 76, 342. (f) Liu, W.; Liu, J.; Ogawa, D.; Nishihara, Y.; Guo, X.; Li, Z. Org. Lett. 2011, 13, 6272. (g) Tian, J.-S.; Loh, T.-P. Angew. Chem., Int. Ed. 2010, 49, 8417. (h) Sundararaju, B.; Tang, Z.; Achard, M.; Sharma, G. V. M; Toupet, L.; Bruneau, C. Adv. Synth. Catal. 2010, 352, 3141. (i) Xia, X.-F.; Shu, X.-Z.; Ji, K.-G.; Yang, Y.-F.; Shaukat, A.; Liu, X.-Y.; Liang, Y.-M. J. Org. Chem. 2010, 75, 2893. (j) Shen, J.; Cai, D.; Kuai, C.; Liu, Y.; Wei, M.; Cheng, G.; Cui, X. J. Org. Chem. 2015, 80, 6584.

(11) (a) Chen, W.; Kang, Y. K.; Wilde, R. G.; Seidel, D. Angew. Chem., Int. Ed. 2014, 53, 5179. (b) Manjappa, K. B.; Jhang, W.-F.; Huang, S.-Y.; Yang, D.-Y. Org. Lett. 2014, 16, 5690.

(12) Haldar, S.; Mahato, S.; Jana, C. K. Asian J. Org. Chem. 2014, 3, 44.

(13) Other N-heterocycles failed to provide the desired imines. However, the corresponding N-benzyl amines were isolated (see the Supporting Information).

(14) Wolken, W. A. M.; Tramper, J.; van der Werf, M. J. Flavour Fragrance J. 2004, 19, 115.

[\(15\) \(a\) Hayashi, Y.;](#page-2-0) Samanta, S.; Gotoh, H.; Ishikawa, H. Angew. Chem., Int. Ed. 2008, 47, 6634. (b) Gotoh, H.; Hayashi, Y. Org. Lett. 2007, 9, 2859. (c) Chan, A.; Scheidt, K. A. Org. Lett. 2005, 7, 905.

(16) Surburg, H.; Panten, J. Common Fragrance and Flavor Materials: Preparation, Properties and Uses; Wiley-VCH: Weinheim, 2006.

(17) (a) Vermoortele, F.; Ameloot, R.; Vimont, A.; Serre, C.; De Vos, D. Chem. Commun. 2011, 47, 1521. (b) Climent, M. J.; Corma, A.; Fornés, V.; Guil- Lopez, R.; Iborra, S. Adv. Synth. Catal. 2002, 344, 1090.

(18) (a) Gunatilaka, A. A. L.; Kingston, D. G. I.; Wijeratne, E. M. K.; Bandara, B. M. R.; Hofmann, G. A.; Johnson, R. K. J. Nat. Prod. 1994, 57, 518. (c) Bandara, B. M. R.; Gunatilaka, A. A. L.; Wijeratne, E. M. K.; MacLeod, J. K. Phytochemistry 1990, 29, 297.

(19) Majumdar, N.; Korthals, K. A.; Wulff, W. D. J. Am. Chem. Soc. 2012, 134, 1357.

(20) (a) Ueda, H.; Yoshida, K.; Tokuyama, H. Org. Lett. 2014, 16, 4194. (b) East, G. C.; Lupton, C. J.; Truter, E. V. Anal. Chim. Acta 1975, 75, 468.